

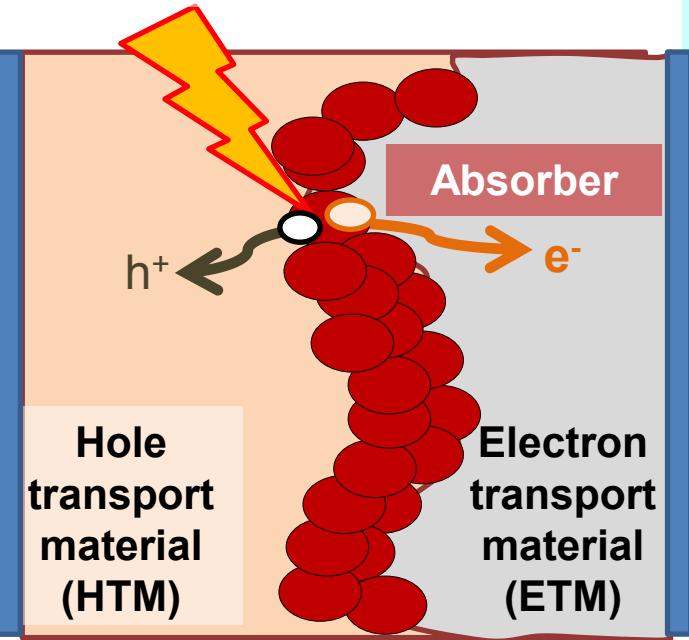
ペロブスカイト型の電荷ダイナミクス

沈 青

電気通信大学大学院理工学研究科 先進理工学専攻

Organic-Inorganic Hybrid Solar Cells

Solar light



Schematic illustration of structure of an organic-inorganic hybrid solar cell (OIHSC)

● Absorber

(dye, quantum dots (QDs), perovskite)

Efficiency of OIHSCs with different absorbers:

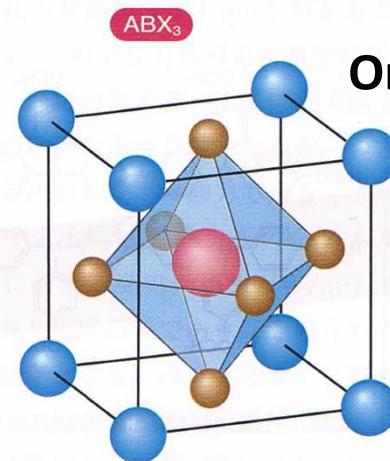
Dye: 1% (1998)¹⁾ → 7.2% (2011)²⁾

Improvements in absorption of dye and conductivity of HTM

Quantum dots (QDs): 5% (2010)³⁾ → 6.3% (2012)⁴⁾

Sb_2S_3 and improvement in HTM

Perovskite lead iodide: 9.7% (2012)⁵⁾ → 20.1% (2014)⁶⁾



Organohalide Lead Perovskites

● : Halogen anion

● : CH_3NH_3^+

● : Pb^{2+}

1) M. Gratzel et al., Nature 1998, 395, 583–585.

2) M. Gratzel et al., J. Am. Chem. Soc. 2011, 133, 18042–18045.

3) S. I. Seok et al., Nano Lett. 2010, 10, 2609–2612.

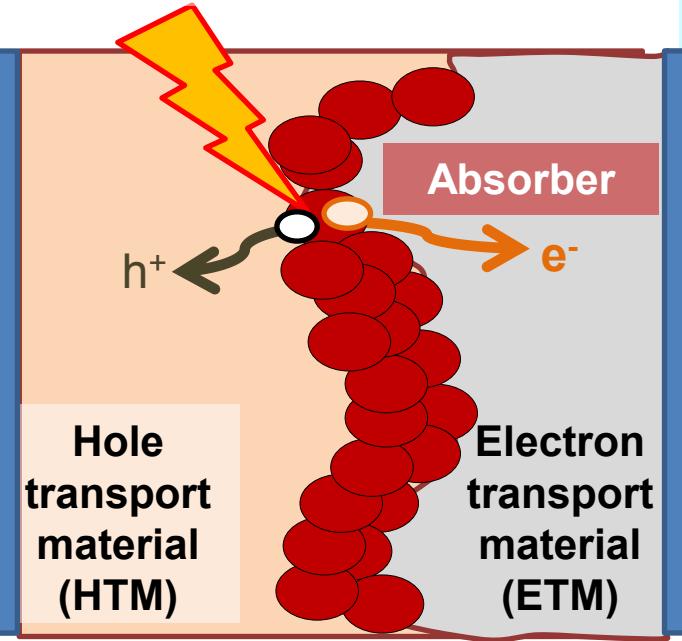
4) S. I. Seok et al., Nano Lett. 2012, 12, 1863–1867.

5) N. G. Park et al., Sci. Rep. 2012, 2, 591.

6) <http://www.nrel.gov/ncpv/>

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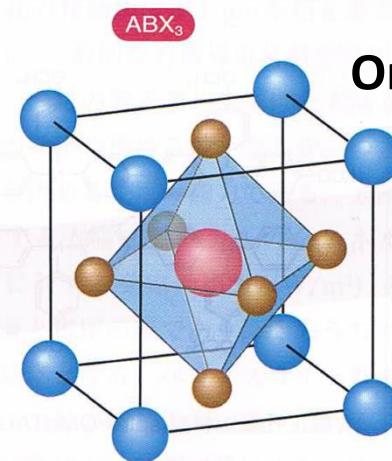
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Organohalide Lead Perovskites

- : Halogen anion
- : CH₃NH₃⁺
- : Pb²⁺

Why the efficiency of perovskite-based OIHSCs can be so high?

Unique properties of Pb perovskite for solid state solar cells

(1) high optical absorption coefficient

N719: $\alpha = 1500 \text{ cm}^{-1}$ at 540 nm; $\text{CH}_3\text{NH}_3\text{PbI}_3$: $\alpha = 15000 \text{ cm}^{-1}$ at 550 nm

(2) long electron and hole diffusion length in the perovskite

$\text{CH}_3\text{NH}_3\text{PbI}_3$: 100 nm; $\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$: > 1000 nm

S. D. Stranks et al., *Science* 342, 341 (2013); G. C. Xing et al., *Science* 342, 344 (2013).

longer lifetimes of photoexcited carriers in Pb perovskite

Unique properties of Pb perovskite for solid state solar cells

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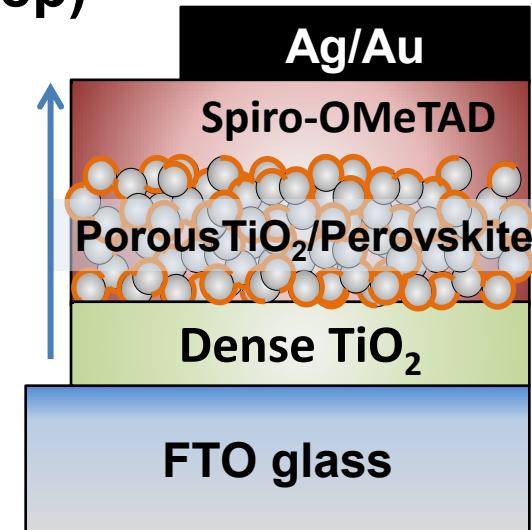
Some open questions for perovskite-based solar cells

1. What is the time scale of charge separation and recombination and which interface do they occur?
2. How about charge separation and charge collection efficiency?
3. How about surface engineering (surface passivation) influence charge separation and collection efficiency?

Our studies on Charge Separation and Recombination in **Perovskite** Solar cells

(1) $\text{TiO}_2/\text{CH}_3\text{NH}_3\text{PbI}_{x}\text{Cl}_{3-x}\text{I}/\text{Spiro-OmeTAD}$ (one step)

- ◊ surface passivation of TiO_2 surface with Y_2O_3
ChemPhysChem, Vol. 15, 1062 (2014).
- ◊ Dependence on TiO_2 morphology
Phys. Chem. Chem. Phys., Vol. 16, 19984 (2014).



(2) $\text{TiO}_2/\text{CH}_3\text{NH}_3\text{PbI}_3/\text{Spiro-OmeTAD}$ (two step)

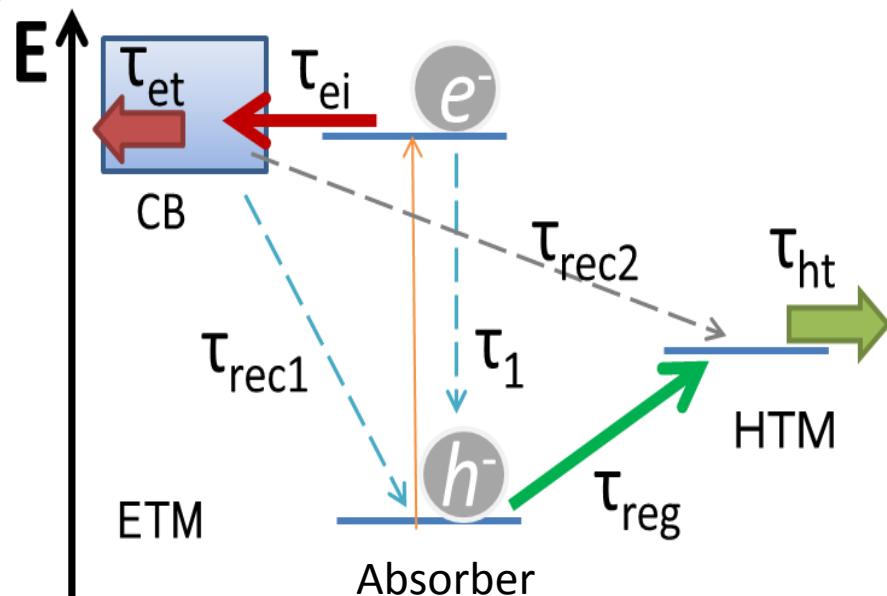
- ◊ Interface engineering by inserting HOCO-RNH₃⁺I⁻ anchor group at Perovskite/TiO₂ interfaces
J. Phys. Chem. C, Vol. 118, 16651 (2014).

(3) $\text{TiO}_2/\text{CH}_3\text{NH}_3\text{Sn}_x\text{Pb}_{1-x}\text{I}_3/\text{Spiro-OmeTAD}$ (one step)

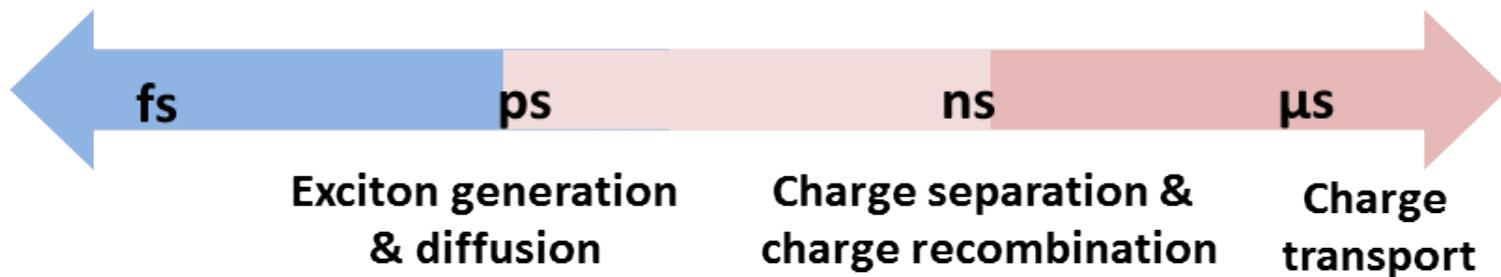
- ◊ Carrier dynamics depends greatly on the ratio x of Sn and Pb
Submitted for publication

Charge Separation and Recombination in OIHSC

One Key for Improving the Efficiency

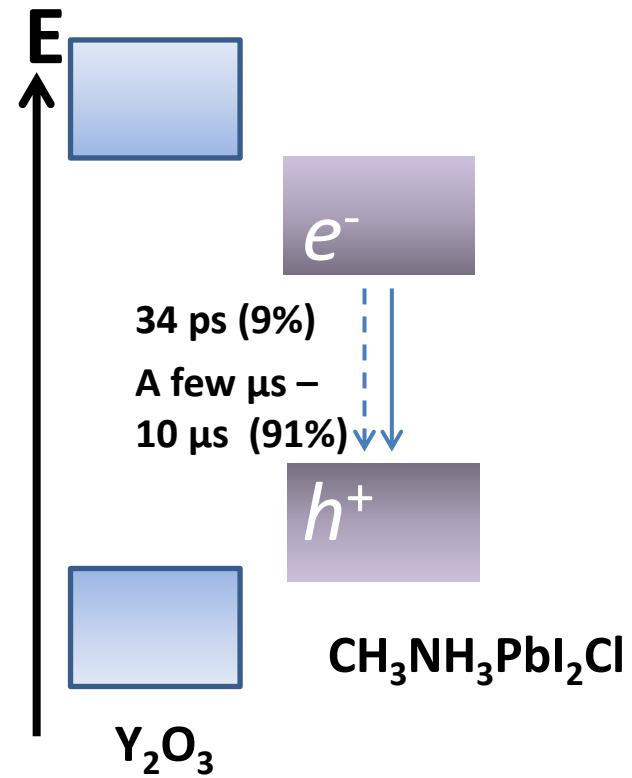
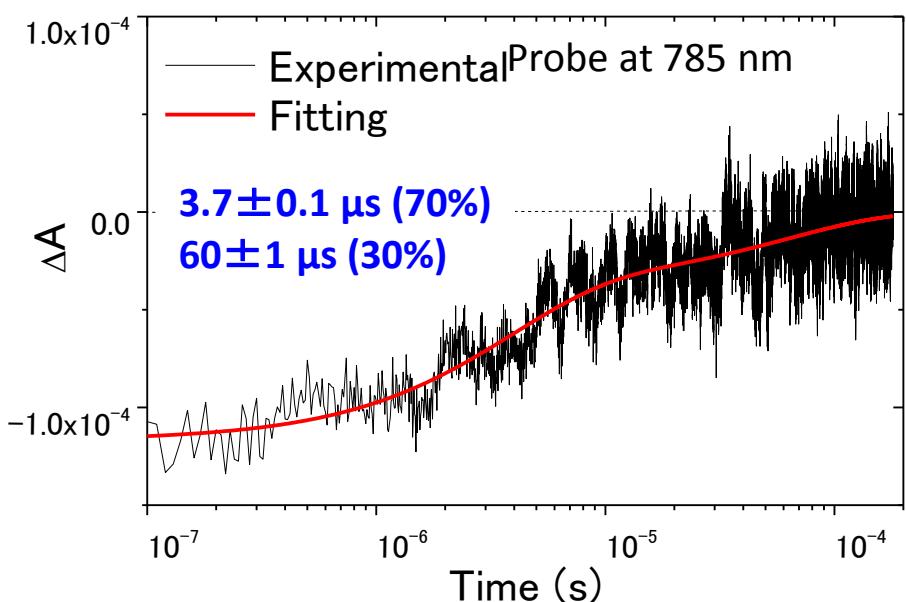
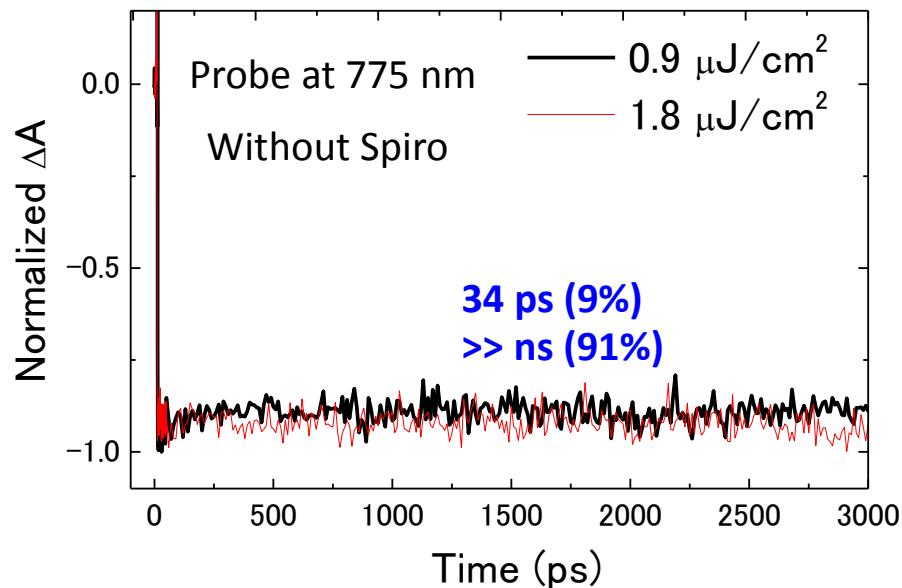


τ_{ei} : electron injection time;
 τ_{reg} : hole transfer time
 τ_1 : life time of photoexcited electron-hole pairs
 τ_{rec1} : recombination time of electrons in TiO_2 with holes in sensitizer
 τ_{rec2} : recombination time of electrons in TiO_2 with holes in HTM
 τ_{et} : electron transport time in ETM
 τ_{ht} : hole transport time in HTM



The charge separation and recombination dynamics from sub-picoseconds to milliseconds can be detected with a transient absorption (TA) method.

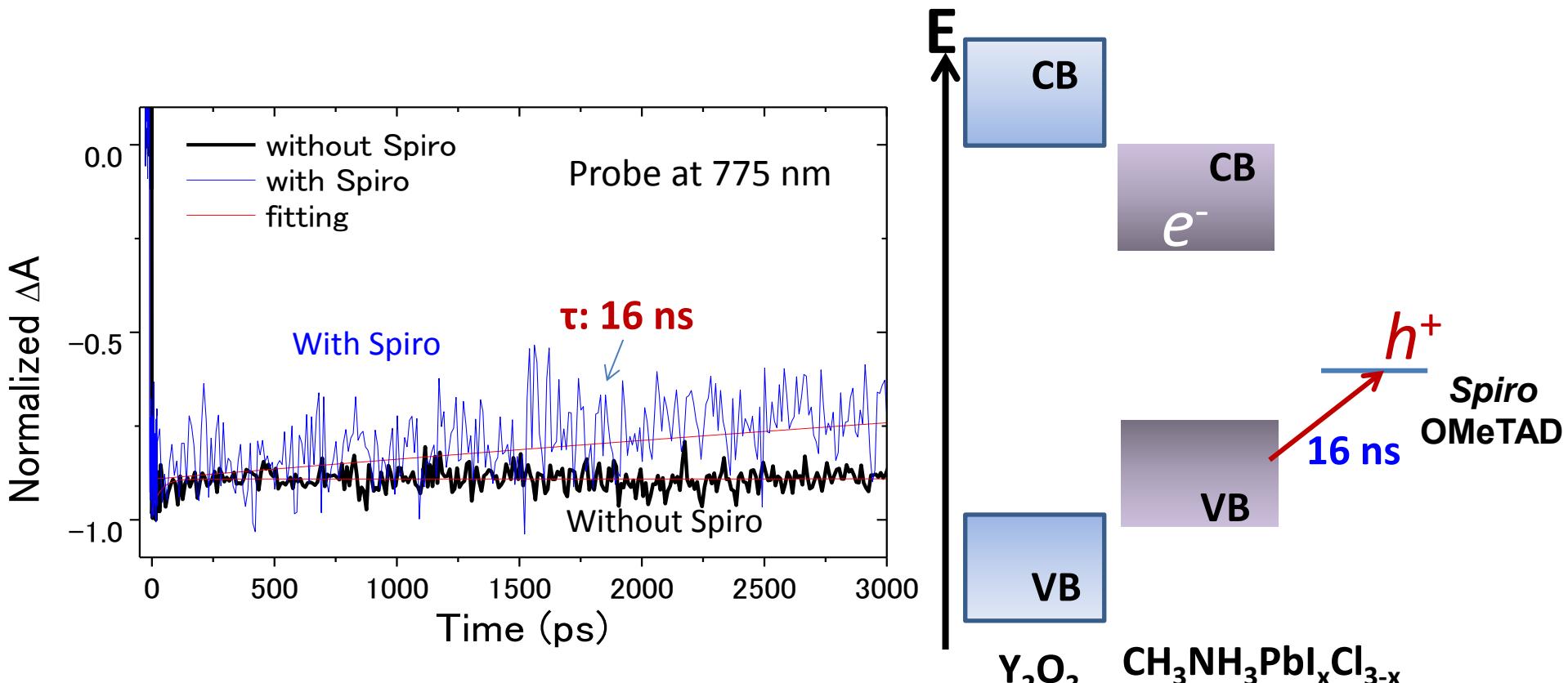
Recombination in $\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}$ on Y_2O_3



Three recombination processes:

- (1) 34 ps (9%)
- (2) 3.7 μs (64%)
- (3) 60 μs (27%)

Charge separation at $\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}/\text{Spiro-OMeTAD}$ interface

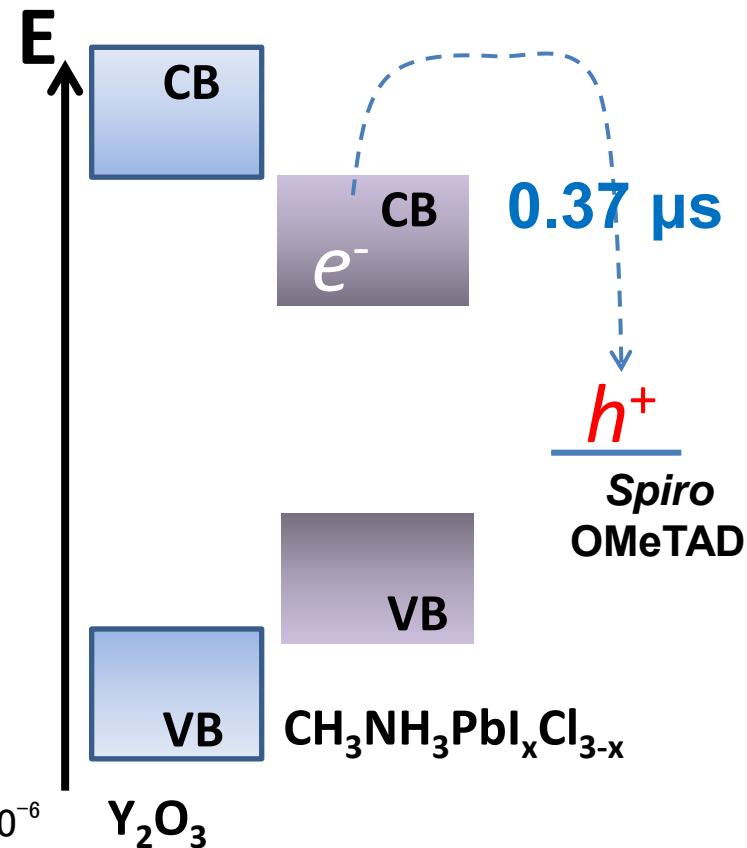
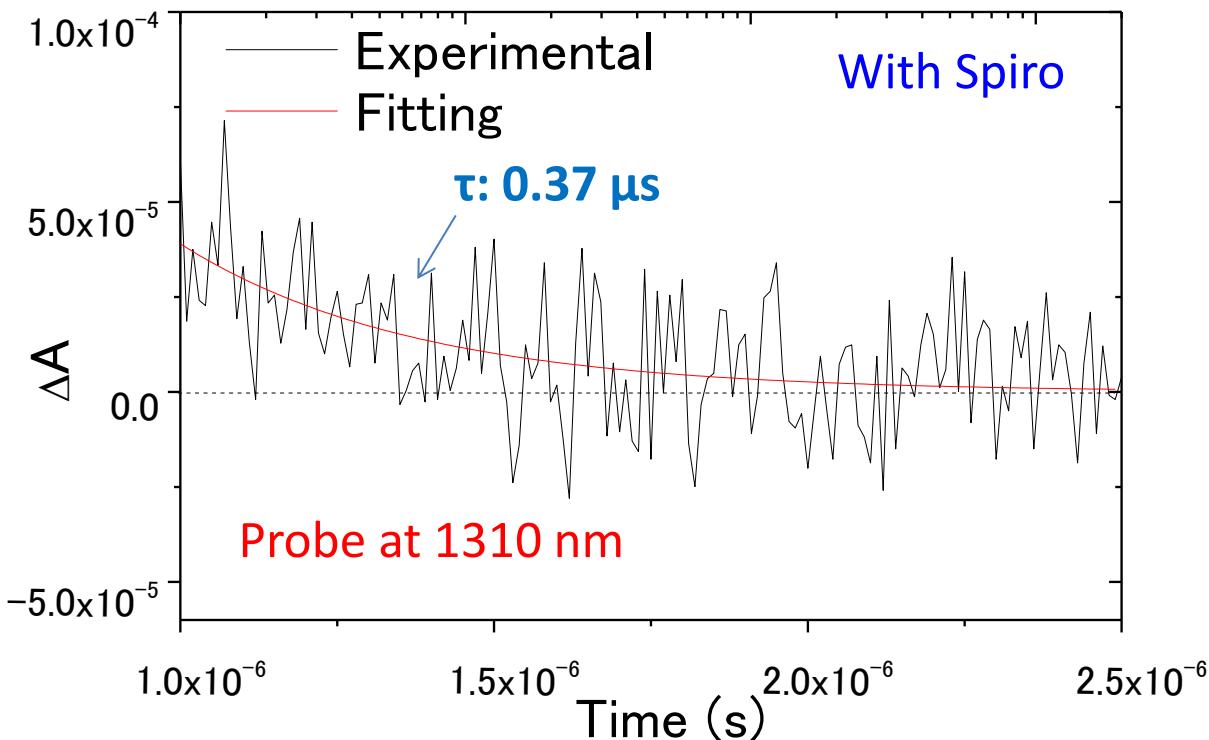


There is charge separation at
 $\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}/\text{Spiro-OMeTAD}$ interface

Hole injection time: 16 ns

charge recombination at $\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}$ /Spiro-OMeTAD interface

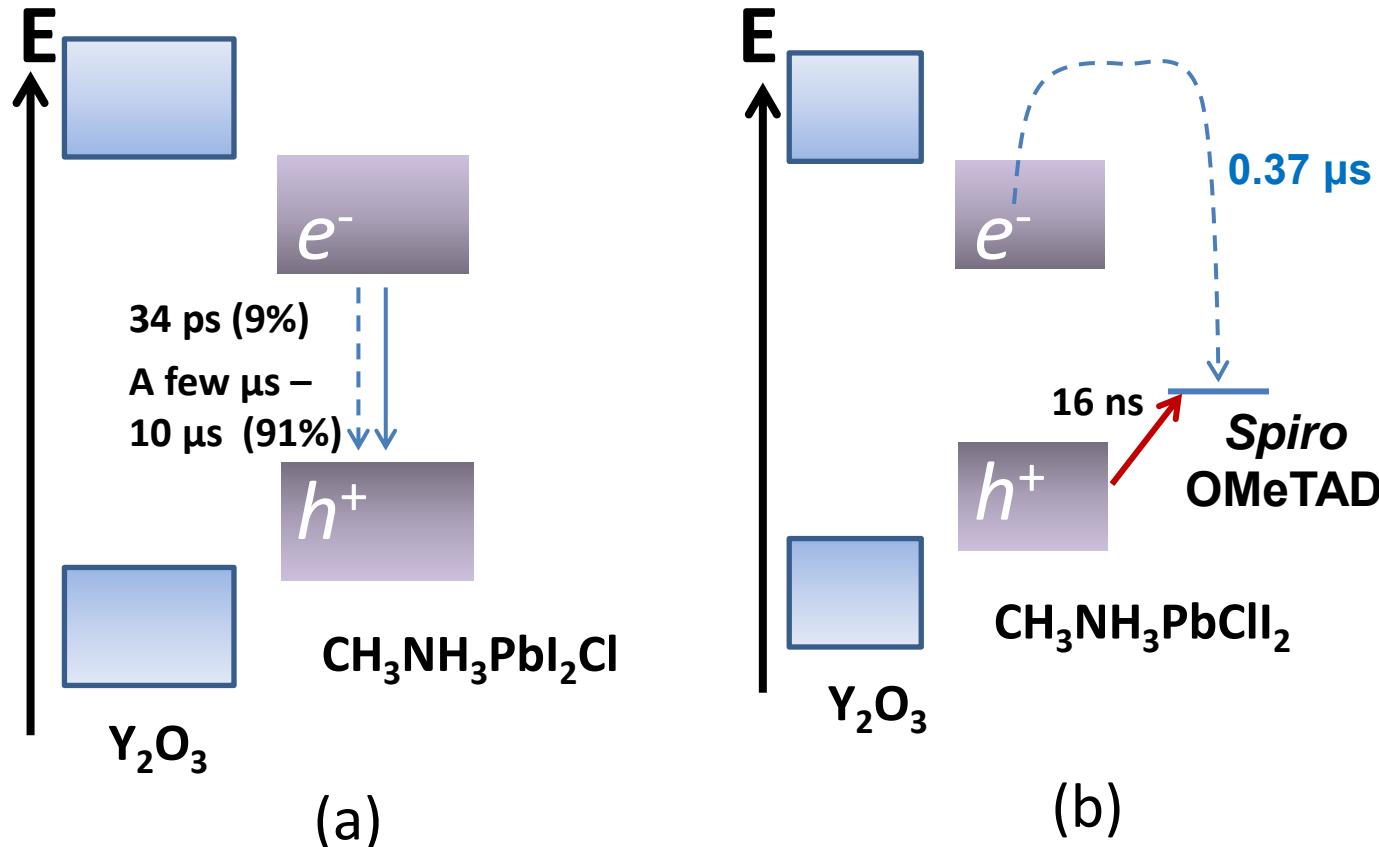
Probe light : 1310 nm → Monitor holes in Spiro



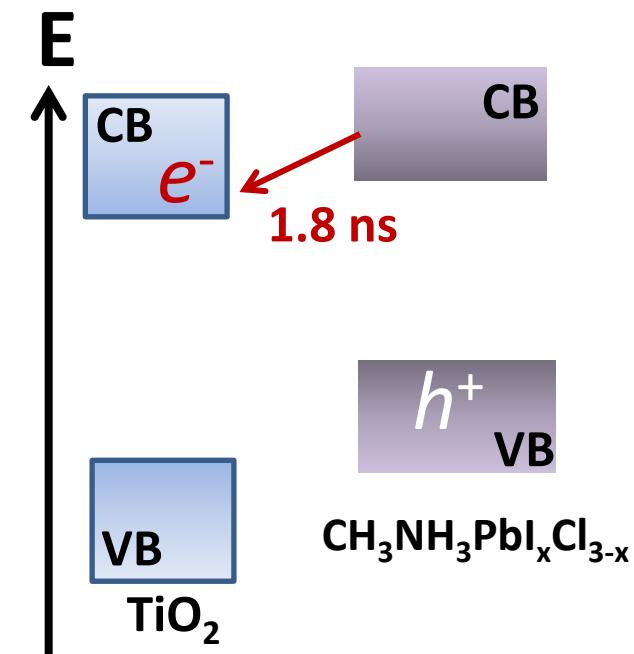
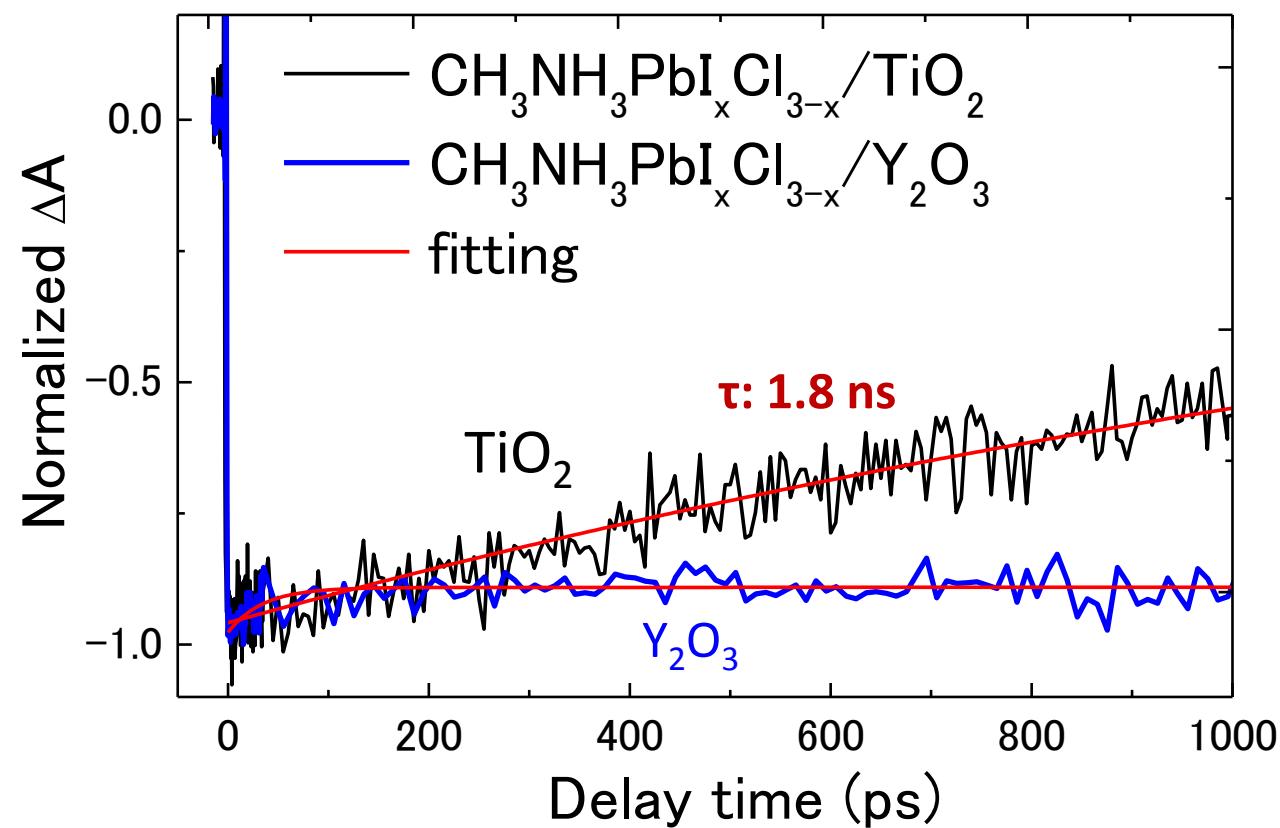
There is charge separation at $\text{CH}_3\text{NH}_3\text{PbClI}_2$ / Spiro-OMeTAD interface

Recombination time: 0.37 μs

Charge separation and recombination at $\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}/\text{Spiro-OMeTAD}$ interface on Y_2O_3



Charge separation at $\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}/\text{TiO}_2$ interface

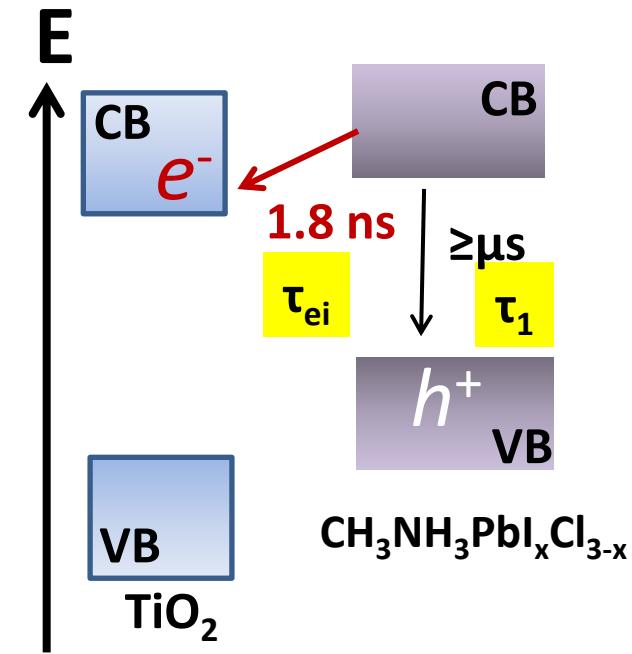
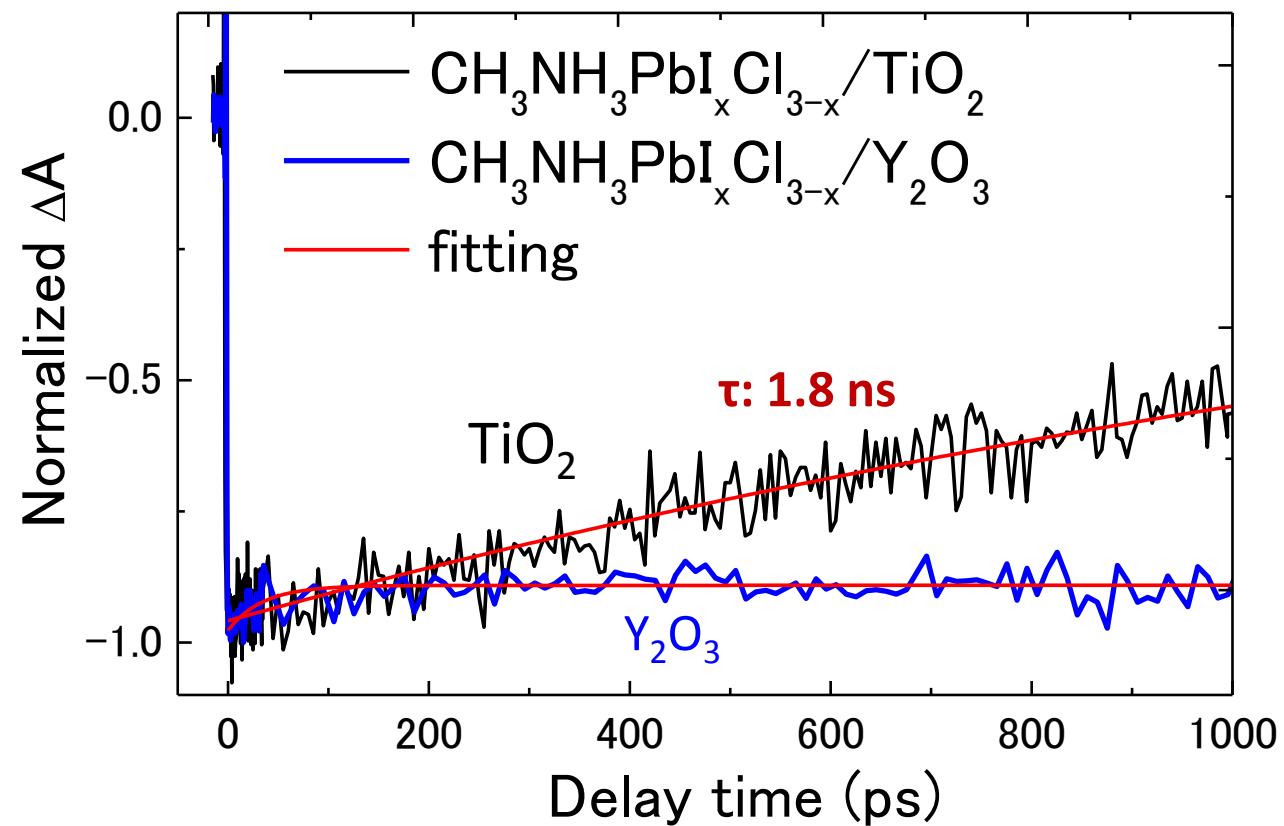


There is electron injection from $\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}$ to TiO_2 .

ChemPhysChem, Vol. 15, 1062 (2014).

Phys. Chem. Chem. Phys., Vol. 16, 19984 (2014).

Charge separation at $\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}/\text{TiO}_2$ interface



Electron injection efficiency η_{ET}

Phys. Chem. Chem. Phys., Vol. 16, 19984 (2014).

$$\eta_{ET} = (1/\tau_{ei}) / (1/\tau_{ei} + 1/\tau_1)$$

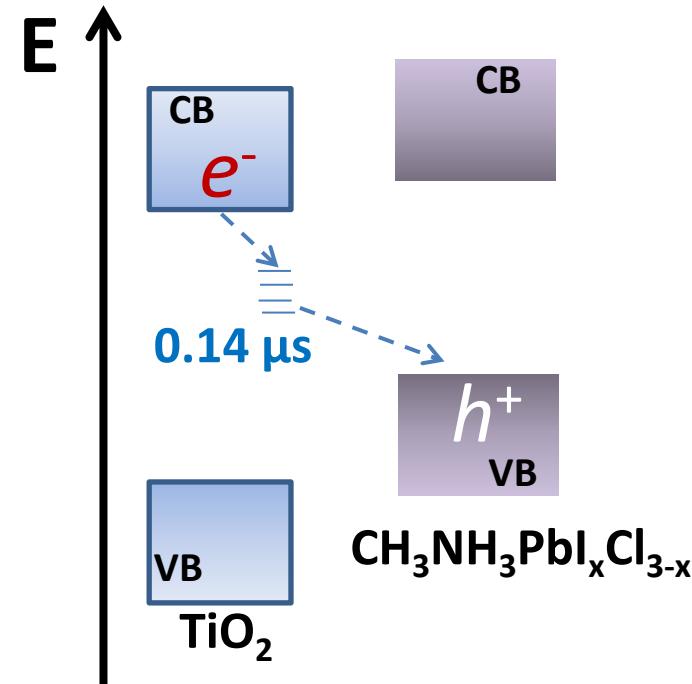
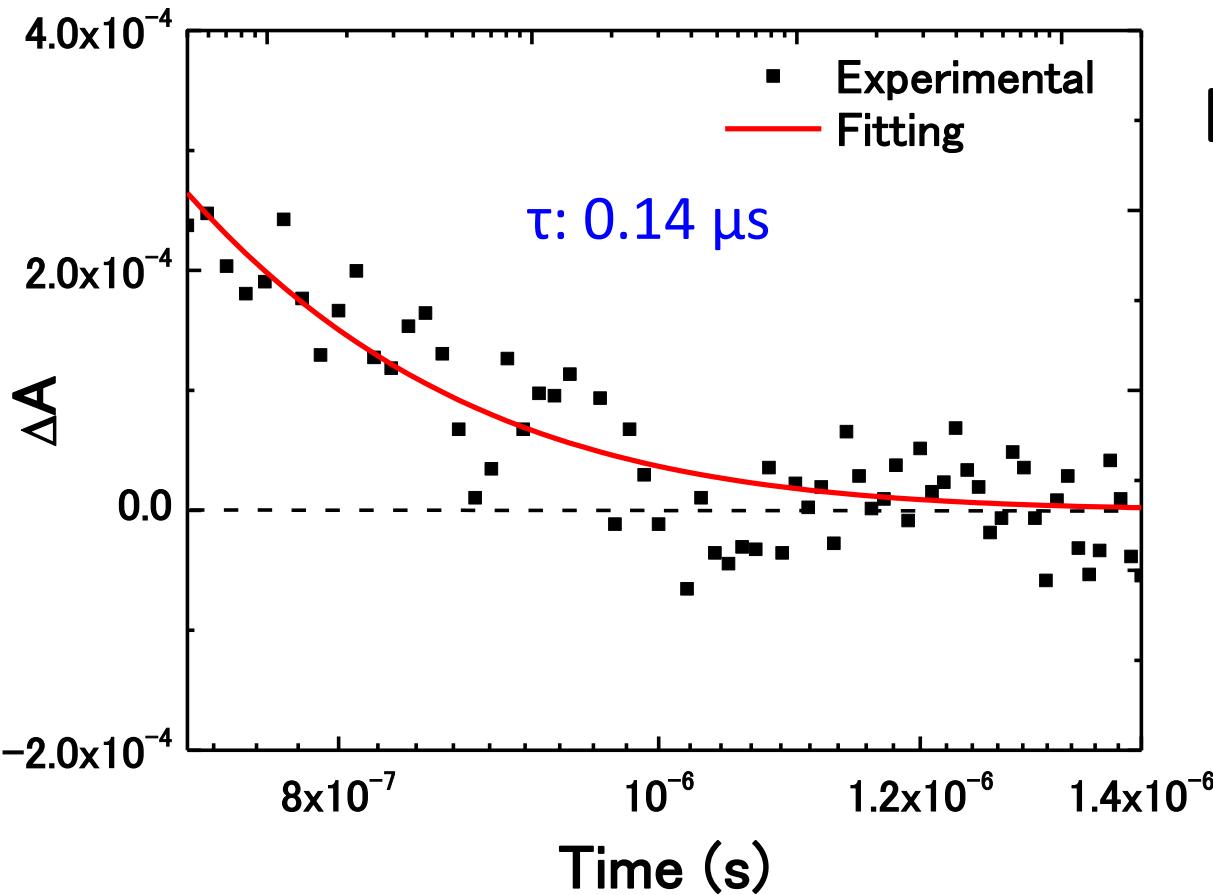
η_{ET} : close to 100%

Charge recombination at $\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}/\text{TiO}_2$ interface

Probe light wavelength: 658 nm → Monitor electrons in TiO_2

T. Yoshihara et al., *The Journal of Physical Chemistry B*, 2004, **108**, 3817-3823.

Physical Chemistry Chemical Physics, 2014, **16**, 5242; 2013, **15**, 11006; 2014, **16**, 5774.

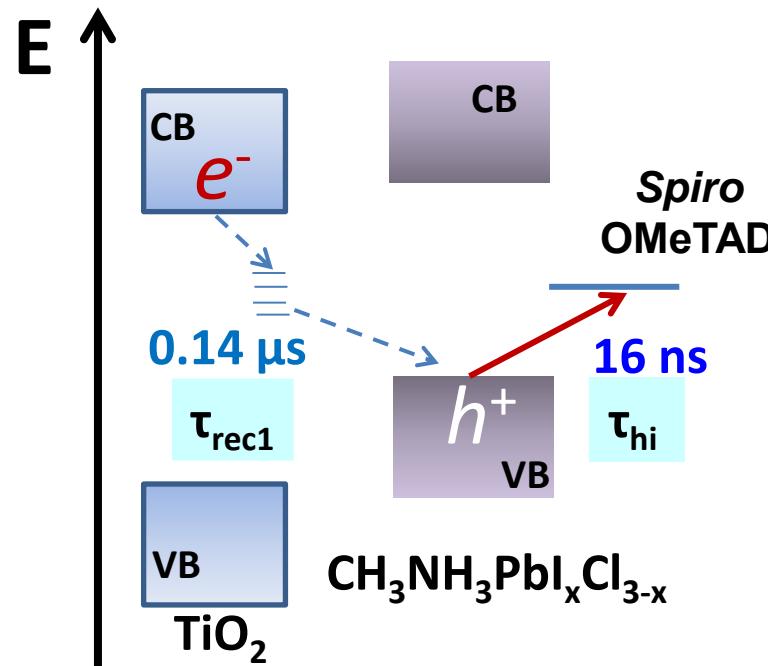
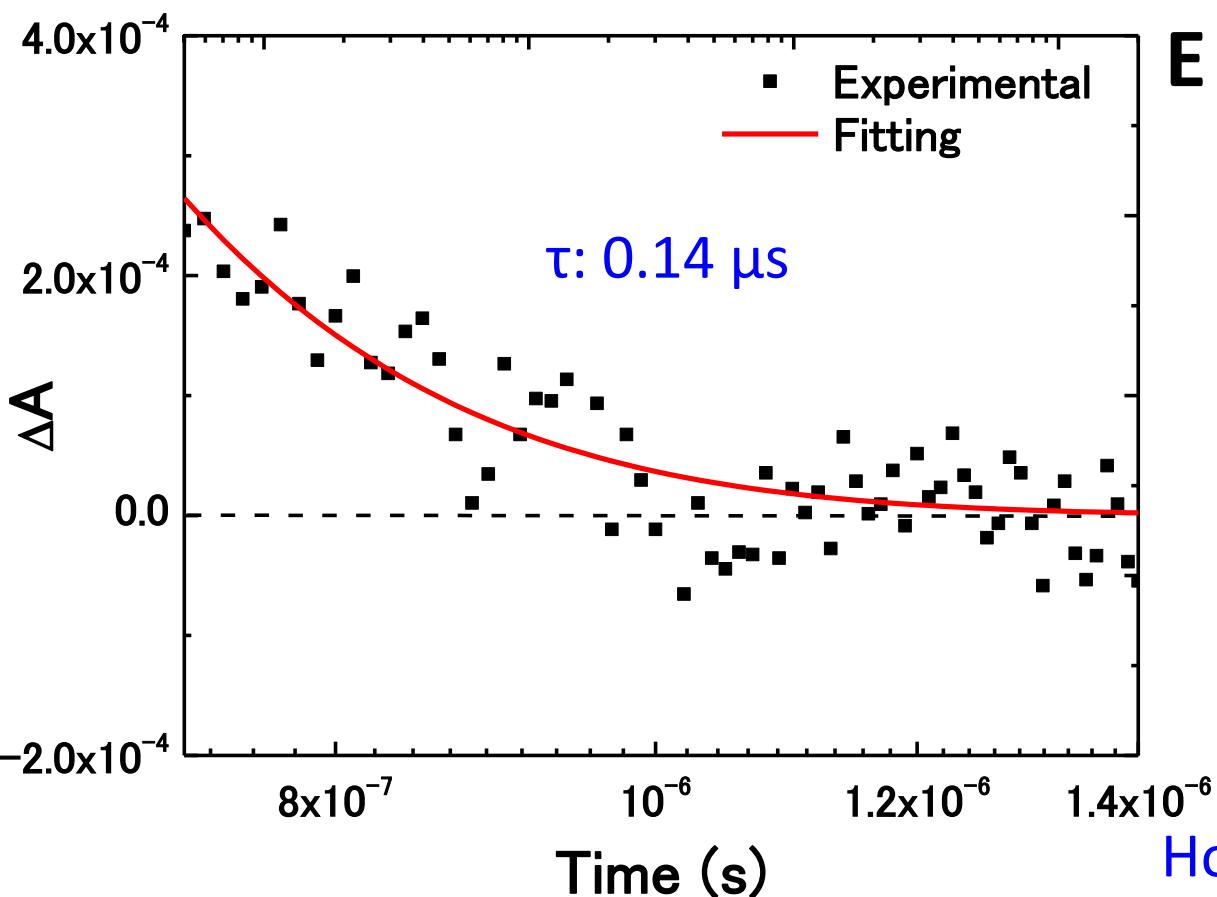


ChemPhysChem, Vol. 15, 1062 (2014).

Phys. Chem. Chem. Phys. (2014). DOI: 10.1039/C4CP03073G.

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Hole injection efficiency η_{HT}

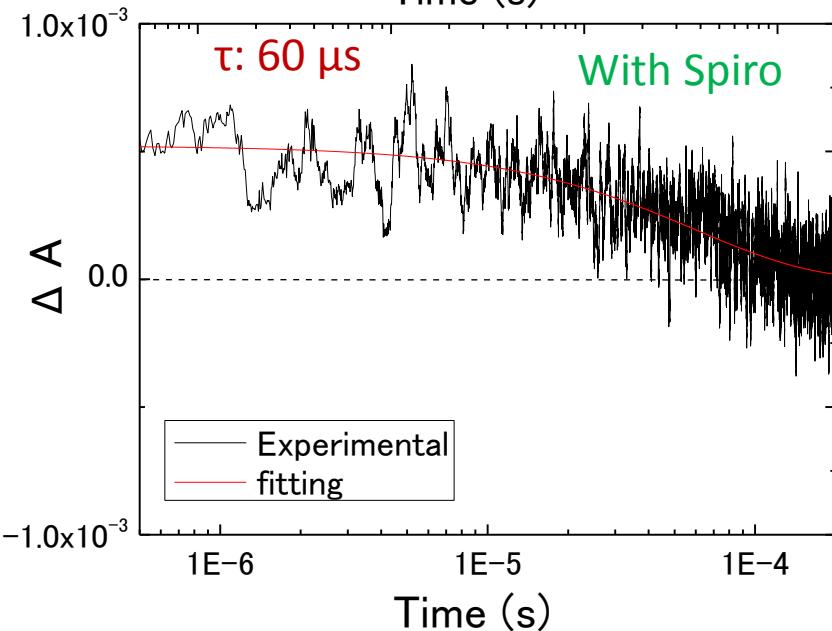
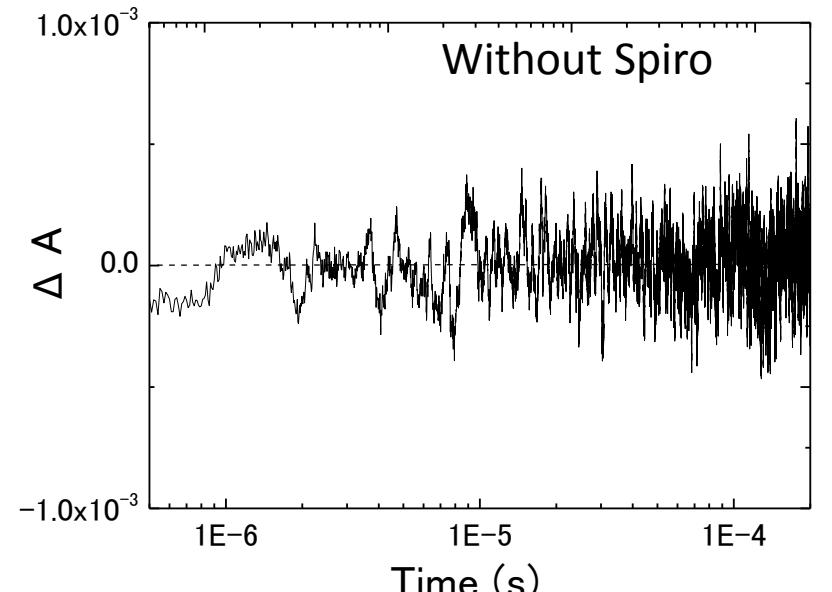
$$\eta_{HT} = (1/\tau_{hi})/(1/\tau_{hi} + 1/\tau_{rec1})$$

η_{HT} : about 90%

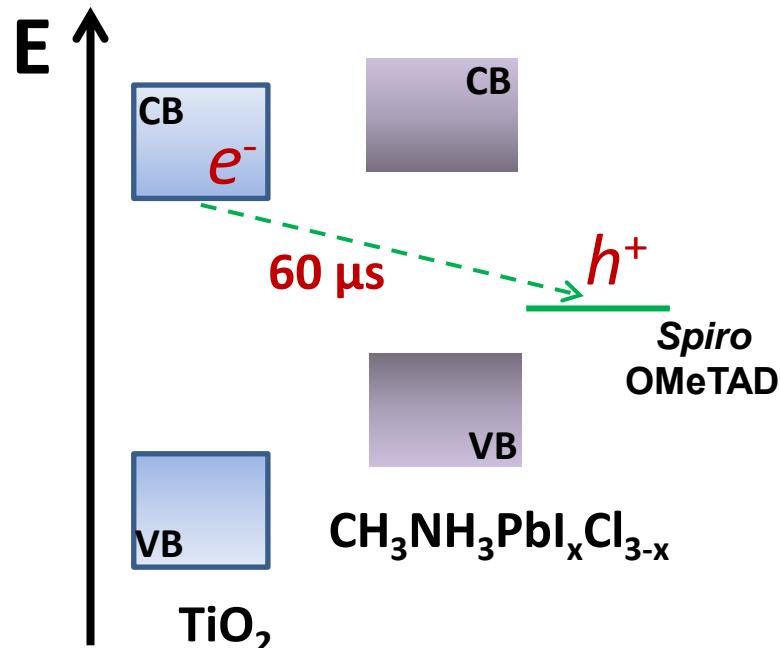
Charge separation efficiency: $\eta_{CS} = \eta_{ET} \eta_{HT} \approx 90\%$

Charge recombination at $\text{TiO}_2/\text{Spiro-OMeTAD}$ interface

Probe light wavelength: 1310 nm → Monitor lifetime of holes in Spiro



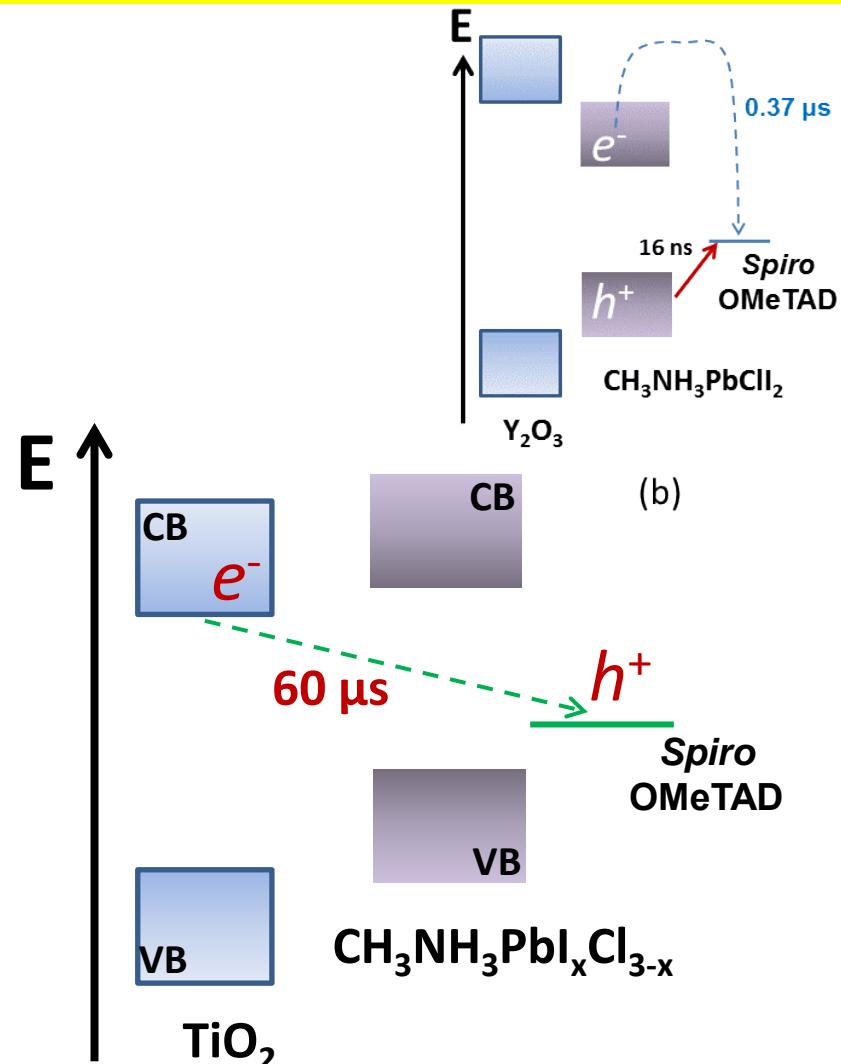
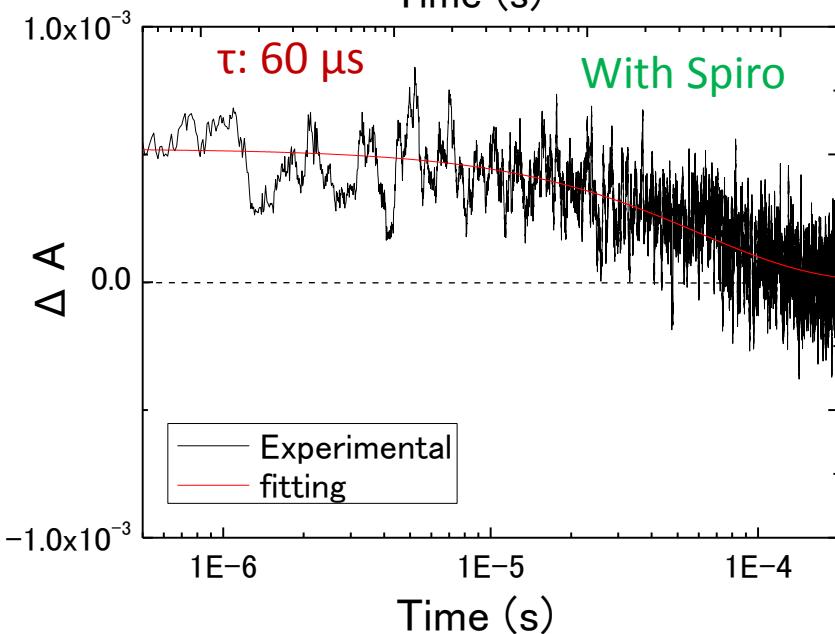
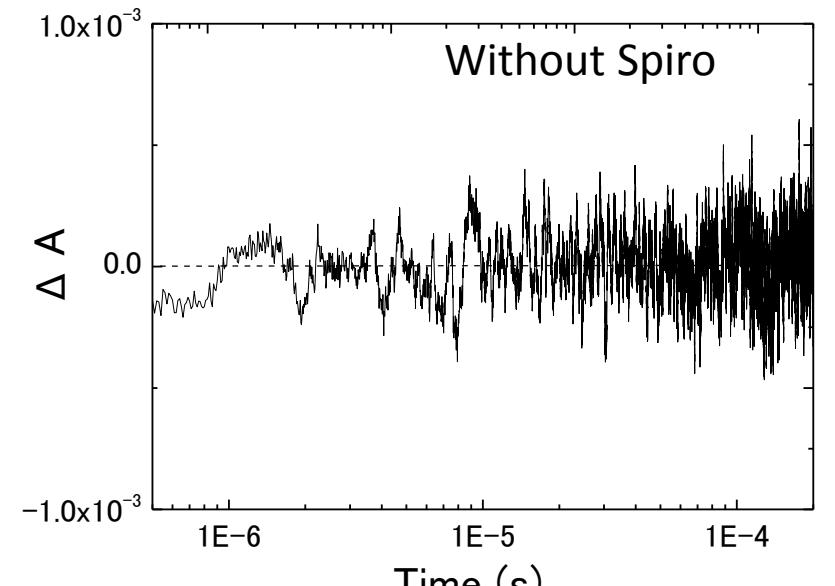
R. Plass et al., *J. Physical Chemistry B*,
2002, **106**, 7578.



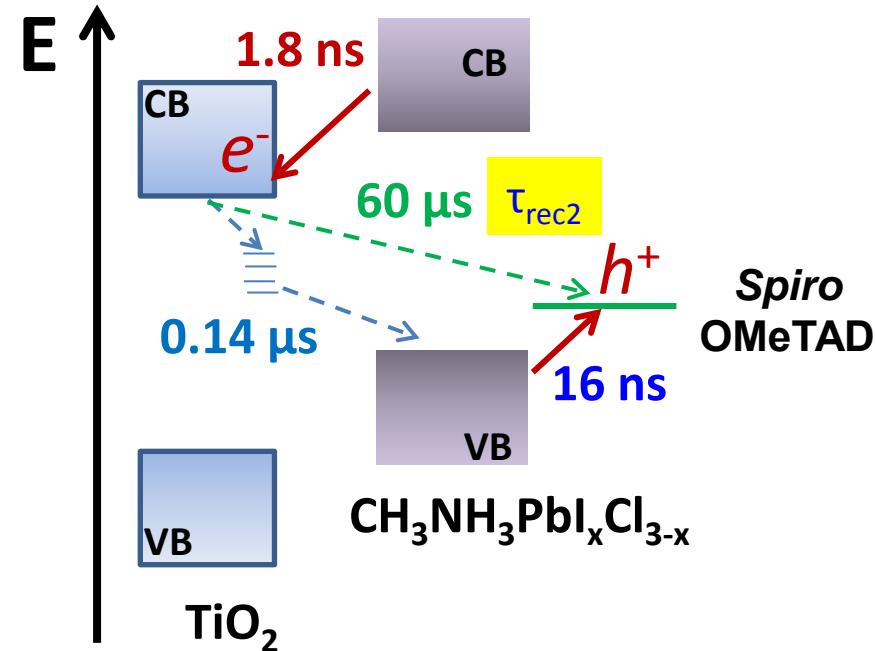
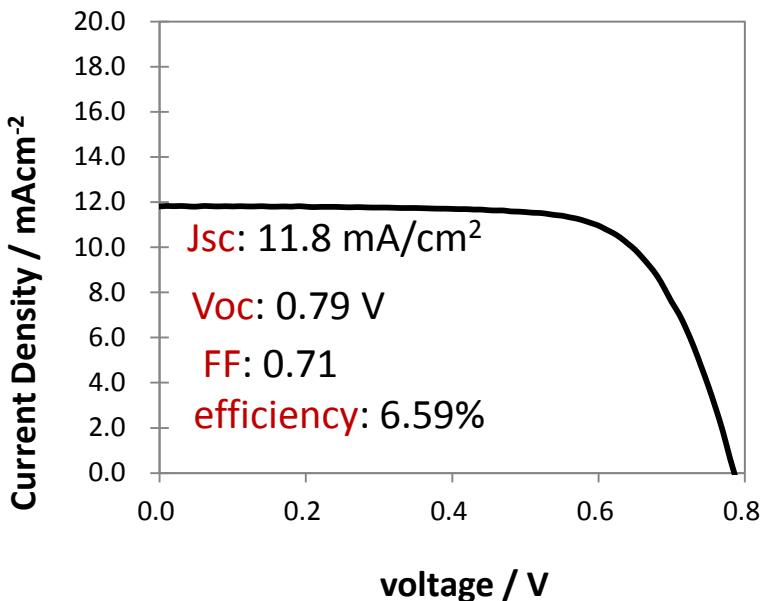
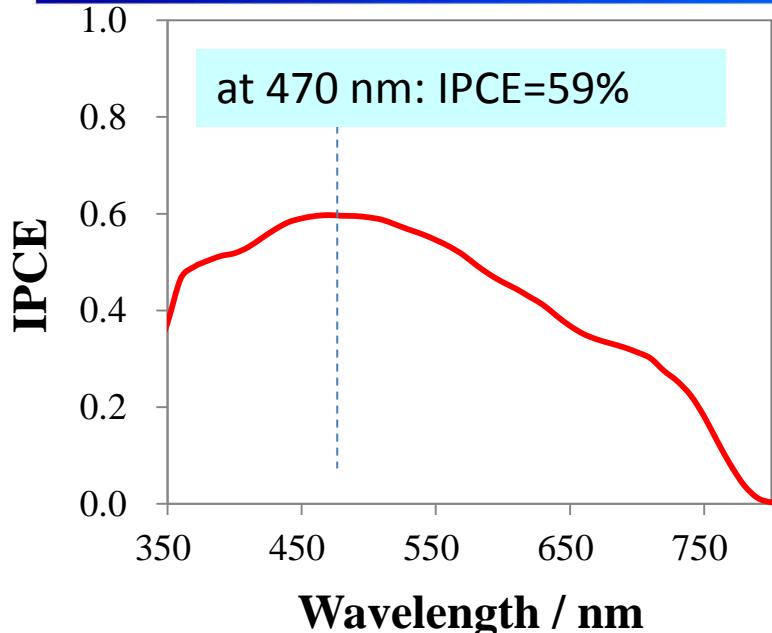
Phys. Chem. Chem. Phys., Vol. 16, 19984 (2014).

Charge recombination at $\text{TiO}_2/\text{Spiro-OMeTAD}$ interface

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Relationship between Carrier Dynamics and Photovoltaic Performance



$$IPCE = \eta_{\text{absorption}} \eta_{\text{CS}} \eta_{\text{CC}}$$

$\eta_{\text{absorption}} \approx 95\%$

$\eta_{\text{CS}} = \eta_{\text{ET}} \eta_{\text{HT}} \approx 90\%$

$\eta_{\text{CC}} < 70\%$

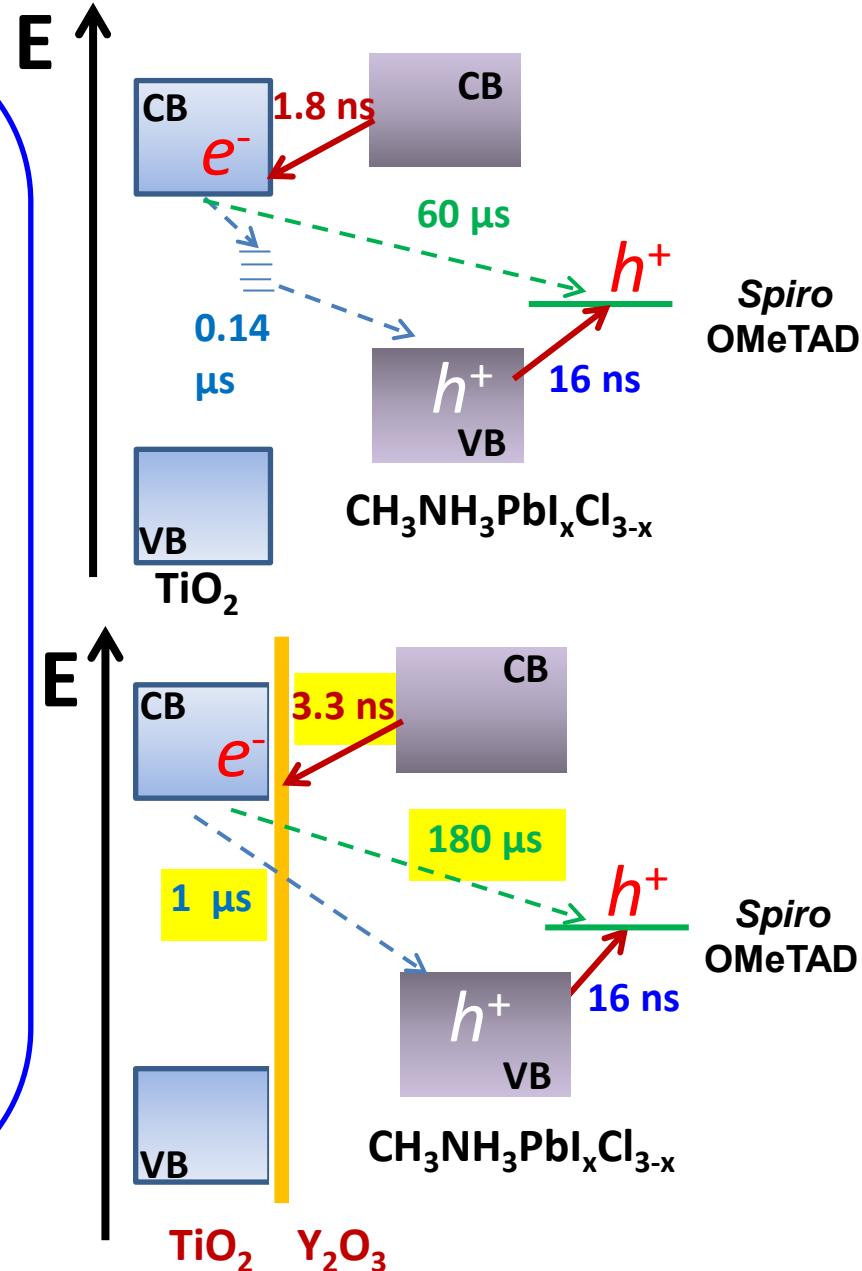
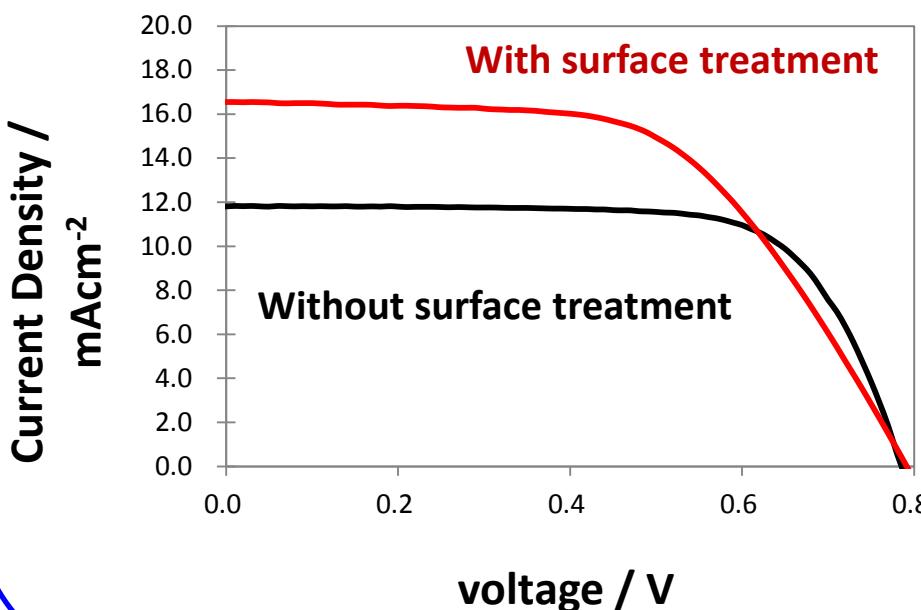
The problem: smaller $\tau_{rec2} \rightarrow$ smaller η_{CC} .

Effects of surface treatment on carrier dynamics and photovoltaic performance

After surface treatment of TiO_2

$\tau_{\text{rec}2}$: $60 \mu\text{s} \rightarrow 180 \mu\text{s}$

J_{sc} : $11.8 \text{ mA/cm}^2 \rightarrow 16.5 \text{ mA/cm}^2$;
efficiency: $6.59\% \rightarrow 7.53\%$



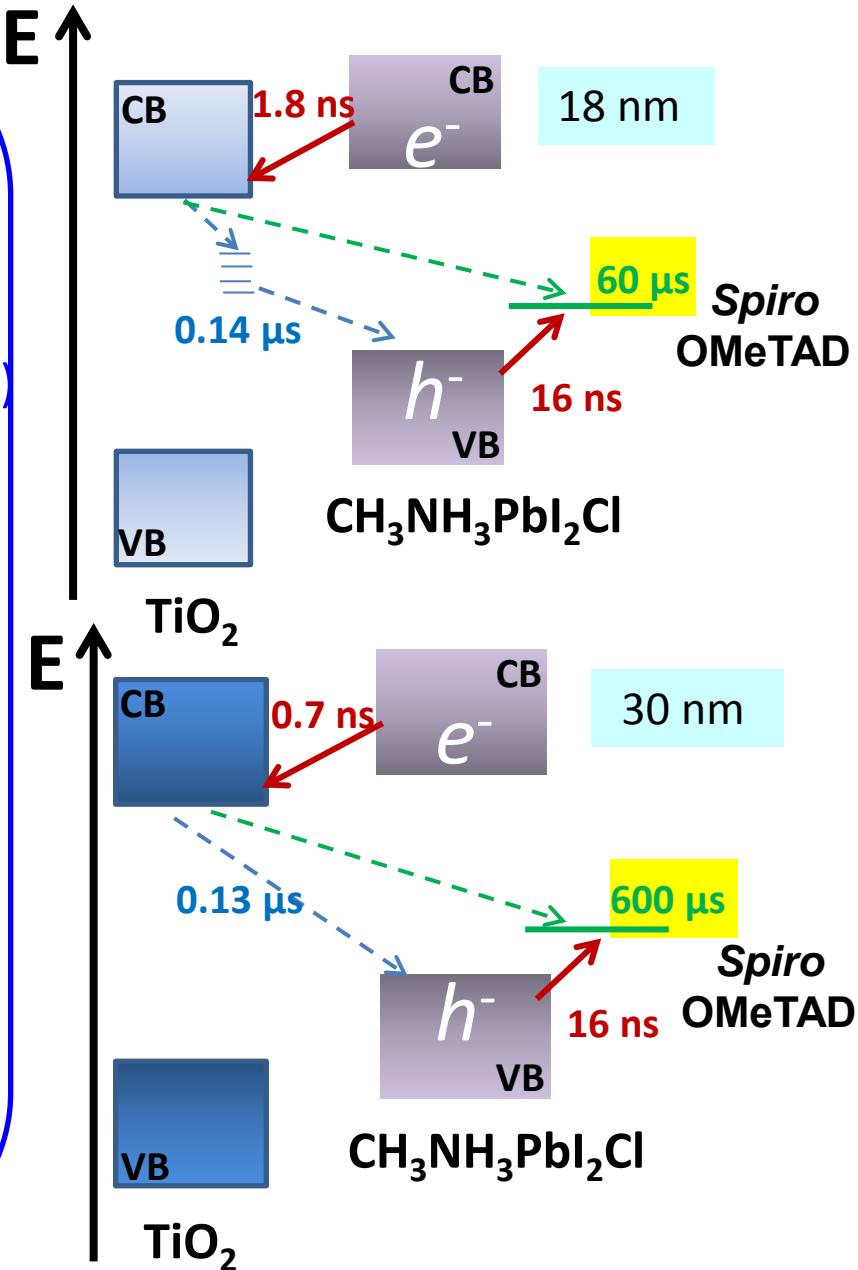
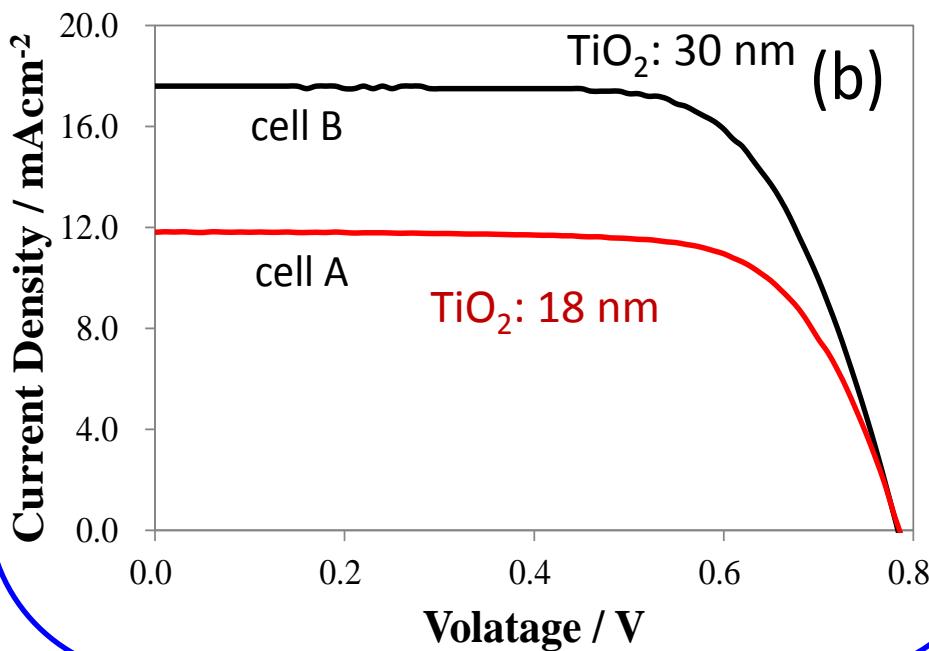
Effects of TiO_2 nanoparticle sizes on carrier dynamics and Photovoltaic Performance

TiO_2 size: 18 nm \rightarrow 30 nm

$\tau_{\text{rec}2}$: 60 μs \rightarrow 600 μs

IPCE(at 470nm): 59% \rightarrow 85% ($\eta_{\text{CC}} \approx 100\%$)

J_{sc} : 11.8 mA/cm 2 \rightarrow 17.6 mA/cm 2 ;
efficiency: 6.59% \rightarrow 9.54%



Summary

- (1) The carrier lifetime of perovskite $\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}$ is as large as μs .
- (2) Charge separation efficiency is as high as 90% for $\text{TiO}_2/\text{CH}_3\text{NH}_3\text{PbI}_x\text{Cl}_{3-x}/\textit{Spiro OMeTAD}$.
- (3) The key factor for improving the efficiency is to decrease the recombination at $\text{TiO}_2/\textit{Spiro OMeTAD}$ interface, which resulted in a smaller charge collection efficiency.



How to suppress the recombination is a key and surface passivation of TiO_2 and interface engineering are important.

Acknowledgment

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九州工業大学：早瀬修二先生、尾込裕平先生

宮崎大学：吉野賢二先生

電通大：豊田太郎先生